

Vidarbha Youth Welfare Society's
INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH
Borgaon (Meghe), Wardha (M.S.)

**MECHANISM OF INTERNAL ASSESSMENT IS
TRANSPARENT AND ROBUST IN TERMS OF
FREQUENCY AND MODE**



R. O. Ganjiwale
(Dr. R. O. Ganjiwale)
Principal
. PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Vidarbha Youth Welfare Society's
INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH
Borgaon (Meghe), Wardha (M.S.)

TABLE OF CONTENT

Sr. No.	Documents	Page number
1.	Examination policy	1
2.	Continuous internal evaluation guidelines	2-4
3.	Internal examination sample schedule	5-6
4.	Internal question paper sample – Theory and Practical	7-8
5.	Internal examination assessment sheet sample - Theory and Practical	9-29
6.	Internal exam attendance sheet - Theory and Practical	30-33
7.	Result sheet sample	34-39
8.	Continuous mode assessment sample	40-45



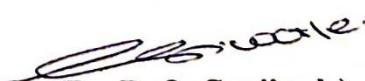
(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Vidarbha Youth Welfare Society's
INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH
Borgaon (Meghe), Wardha (M.S.)

2.5.1 Mechanism of internal assessment is transparent and robust in terms of frequency and mode. Write description within 200 words

The college has tools for internal assessment such as unit test, open book test, and assignment test conducted by individual teachers from time to time, attendance will be monitor for every month as well as before every sessional exam and each time students are informed about their short attendance, students are also assessing on the basis of project, theory sessional examination, practical examination, and viva-voce. The institution ensures transparency in internal assessment by using above mentioned tool for the assessment of the students and the outcomes are communicated to them. The faculty members give personal and academic counselling to the students. For evaluating the overall development of the students, due weightage is given to behavioural aspect, independent learning and communication skills.




(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

CONTINUOUS INTERNAL EVALUATION GUIDELINES

Internal assessment: Continuous mode

The marks allocated for Continuous mode of Internal Assessment shall be awarded as per the scheme given below.

Table-XI: Scheme for awarding internal assessment: Continuous mode

Theory			
Criteria	Maximum Marks		
Attendance (Refer Table – XII)	4	2	
Academic activities (Average of any 3 activities e.g. quiz, assignment, open book test, field work, group discussion and seminar)	3	1.5	
Student – Teacher interaction	3	1.5	
Total	10	5	
Practical			
Attendance (Refer Table – XII)	2		
Based on Practical Records, Regular viva voce, etc.	3		
Total	5		

Table- XII: Guidelines for the allotment of marks for attendance

Percentage of Attendance	Theory	Practical
95 – 100	4	2
90 – 94	3	1.5
85 – 89	2	1
80 – 84	1	0.5
Less than 80	0	0

Sessional Exams

Two Sessional exams shall be conducted for each theory / practical course as per the schedule fixed by the college(s). The scheme of question paper for theory and practical Sessional examinations is given below. The average marks of two Sessional exams shall be computed for internal assessment as per the requirements given in tables – X.

Sessional exam shall be conducted for 30 marks for theory and shall be computed for 15 marks. Similarly Sessional exam for practical shall be conducted for 40 marks and shall be computed for 10 marks.

Question paper pattern for theory Sessional examinations

For subjects having University examination

I. Multiple Choice Questions (MCQs) OR Objective Type Questions (5 x 2) (Answer all the questions)	= $10 \times 1 = 10$ OR = $05 \times 2 = 10$
I. Long Answers (Answer 1 out of 2) II. Short Answers (Answer 2 out of 3)	= $1 \times 10 = 10$ = $2 \times 5 = 10$ <hr/> Total = 30 marks

For subjects having Non University Examination

I. Long Answers (Answer 1 out of 2)	=	1 x 10 = 10
II. Short Answers (Answer 4 out of 6)	=	4 x 5 = 20
	Total	= 30 marks

Question paper pattern for practical sessional examinations

I. Synopsis	=	10
II. Experiments	=	25
III. Viva voce	=	05
	Total	= 40 marks

1. Promotion and award of grades

A student shall be declared PASS and eligible for getting grade in a course of B.Pharm.program if he/she secures at least 50% marks in that particular course including internal assessment. For example, to be declared as PASS and to get grade, the student has to secure a minimum of 50 marks for the total of 100 including continuous mode of assessment and end semester theory examination and has to secure a minimum of 25 marks for the total 50 including internal assessment and end semester practical examination.

2. Carry forward of marks

In case a student fails to secure the minimum 50% in any Theory or Practical course as specified in 12, then he/she shall reappear for the end semester examination of that course. However his/her marks of the Internal Assessments shall be carried over and he/she shall be entitled for grade obtained by him/her on passing.

3. Improvement of internal assessment

A student shall have the opportunity to improve his/her performance only once in the Sessional exam component of the internal assessment. The re-conduct of the Sessional exam shall be completed before the commencement of next end semester theory examinations.

4. Re-examination of end semester examinations

Reexamination of end semester examinations shall be conducted as per the schedule given in table XIII. The exact dates of examinations shall be notified from time to time.



(Dr. R. O. Ganjiwale)
Principal

PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Table-XIII: Tentative schedule of end semester examinations

Semester	For Regular Candidates	For Failed Candidates
I, III, V and VII	November / December	May / June
II, IV, VI and VIII	May / June	November / December

Question paper pattern for end semester theory examinations

For 75 marks paper

I. Multiple Choice Questions(MCQs)	=	20 x 1	= 20
OR		OR	
Objective Type Questions (10 x 2)	=	10 x 2	= 20
(Answer all the questions)			
II. Long Answers (Answer 2 out of 3)	=	2 x 10	= 20
III. Short Answers (Answer 7 out of 9)	=	7 x 5	= 35

Total	=	75 marks	

For 50 marks paper

I. Long Answers (Answer 2 out of 3)	=	2 x 10 = 20
II. Short Answers (Answer 6 out of 8)	=	6 x 5 = 30

Total	=	50 marks

For 35 marks paper

I. Long Answers (Answer 1 out of 2)	=	1 x 10 = 10
II. Short Answers (Answer 5 out of 7)	=	5 x 5 = 25

Total	=	35 marks

Question paper pattern for end semester practical examinations

I. Synopsis	=	5
II. Experiments	=	25
III. Viva voce	=	5

Total	=	35 marks




 (Dr. R. O. Ganjiwale)
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe) Wardha
SECOND SESSIONAL EXAMINATION (THEORY), 2023-24
B.PHARM - Sem -VII
***** TIME - TABLE *****

Day & Date	Time	Class	Subject
WEDNESDAY 01/11/2023	11.00 A.M. TO 12.00 P.M.	B. Pharm SEM-VII	PHARMACY PRACTICE
THURSDAY 02/11/2023	11.00 A.M. TO 12.00 P.M.	B. Pharm SEM-VII	INDUSTRIAL PHARMACY
FRIDAY 03/11/2023	11.00 A.M. TO 12.00 P.M.	B. Pharm SEM-VII	NOVEL DRUG DELIVERY SYSTEM
SATURDAY 04/11/2023	11.00 A.M. TO 12.00 P.M.	B. Pharm SEM-VII	INSTRUMENTAL METHODS OF ANALYSIS

(Ms. S. P. Gautam/ Dr. B. R. Gandhare)

Exam-in-Charge

Copy to:-

1. Examination in Charge
2. Notice Board
3. Notice Board File
4. Office Copy

(Dr. R. O. Ganjiwale)

Principal

Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha,



INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe) Wardha

SECOND SESSIONAL PRACTICAL EXAMINATION, 2023-2024

B. PHARM- (SEM-VII)

T I M E - T A B L E

DAY & DATE	TIME	CLASS	BATCH	SUBJECT
MONDAY 09/10/2023	12.30 P.M. TO 04.30 P.M.	Sem-VII	A	INSTRUMENTAL METHODS OF ANALYSIS
TUESDAY 10/10/2023	12.30 P.M. TO 04.30 P.M.	Sem-VII	B	INSTRUMENTAL METHODS OF ANALYSIS
WEDNESDAY 11/10/2023	12.30 P.M. TO 04.30 P.M.	Sem-VII	C	INSTRUMENTAL METHODS OF ANALYSIS

(Ms. S. P. Gautam)

Exam-in-Charge

Copy to:-

1. Examination in Charge
2. Notice Board
3. Notice Board File
4. Office Copy

(Dr. R. O. Ganjiwale)

Principal

Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha,





Institute of Pharmaceutical Education and Research, Wardha
B. Pharm.

SECOND SESSIONAL EXAM-2023 (Question Paper)

Marks: 30
Minutes

Subject: [BP701T] Instrumental Methods of Analysis - Theory

Date: 4 November, 2023

Duration: 60

Note:

1. Attempt all questions from Que. No. 1. Each for 2 marks
2. Attempt Any ONE question from Que. No. 2 for 10 marks
3. Attempt Any TWO questions from Que. No. 3. Each of 5 marks
4. Use of calculators is permitted

Que. No. 1] Attempt All Questions ($5 \times 2 = 10$ marks)

a) Define the terms Capacity factor, Selectivity factor, Retention time and Dead time in chromatography.

Bloom's Level: Evaluate, Apply, Understand, Remember **Course Outcome:** CO3

b) Define the term HETP and explain its corelation with the theoretical plates.

Bloom's Level: Evaluate, Understand, Remember **Course Outcome:** CO3

c) Explain Fermi's Resonance with example.

Bloom's Level: Evaluate, Understand **Course Outcome:** CO1

d) How many Translational, Rotational and Vibrational degrees of freedom are there for:

(i) Linear CO_2 molecule

(ii) Non-linear H_2S molecule

Bloom's Level: Remember **Course Outcome:** CO1, CO2

e) Discuss the 2D technique of development in TLC.

Bloom's Level: Create, Apply, Understand **Course Outcome:** CO4

Que. No. 2] Attempt Any ONE question from the following: ($1 \times 10 = 10$ marks)

a) Sketch well labelled diagram of Gas chromatographic unit. Write in brief on columns and stationary phases. Write note on Thermal Conductivity Detector used in GC. (2 + 5 + 3)

Bloom's Level: Apply, Remember **Course Outcome:** CO5

b) Discuss in brief the various factors affecting the vibrational frequency in IR Spectroscopy.

Bloom's Level: Create, Understand **Course Outcome:** CO1

Que. No. 3] Attempt Any TWO questions from the following: ($2 \times 5 = 10$ marks)

a) Sketch well labelled diagram of HPLC unit, and write in brief on Reciprocating Pump OR Bulk Property Detector. (2 + 3)

Bloom's Level: Apply, Remember **Course Outcome:** CO5

b) Describe various development techniques in Paper Chromatography.

Bloom's Level: Create, Apply, Understand **Course Outcome:** CO4

c) Describe the various factors that affects electrophoretic mobility in Electrophoresis. (5.00)

Marks)

Bloom's Level: Understand **Course Outcome:** CO6



Dr. R. O. Ganjiwale

Principal

PRINCIPAL

**Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha**

*P.M. Erik
27.11.23*

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

SECOND SESSIONAL PRACTICAL EXAMINATION 2023-2024

B. PHARM-IV (SEM-VII)

SUBJECT: Instrumental Method of Analysis (Practical)

Time: 4hrs

Date: 9th, 10th and 11th Oct 2023

Max. Marks: 40

Instruction: 1) Solve all questions

Que.No. 1 Synopsis

(5x2=10M)

Batch 'A'

- a. Explain the principle of HPLC?
- b. Draw schematic diagram of GC?
- c. Differentiate between TLC and HPTLC?
- d. What are some of the applications of Infrared spectroscopy?
- e. Enlist most widely used carrier gases and detectors in Gas chromatography?

Batch 'B'

- a. How does UV- Visible spectroscopy work?
- b. Differentiate between Adsorption and Partition Chromatography?
- c. Enlist factors that affect electrophoretic mobility?
- d. What are the applications of Column chromatography?
- e. What are advantages and applications of HPLC Technique?

Batch 'C'

- a. Explain the principle of HPLC?
- b. Draw schematic diagram of GC?
- c. Write short note on ion exchange chromatography?
- d. Classify different chromatographic technique?
- e. What are the advantages and application of IR spectroscopy?

Q. NO. 2

(25M)

To perform assay of Paracetamol tablet as per IP by Reference Standard Method.

Q. NO. 3

(05M)

Viva-Voce



Dr. R. O. Ganjiwale

Principal

PRINCIPAL

Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Om

Vidarbha Youth Welfare Society's
INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH

Name	Krushnali S. Thakare	Subject:	Instrumental methods of Analysis	Examination First/Second Third/ Be Sessional
Class	B.PH.-IV (SEM-VII)	Examination Seat No.	15	Date 04/11/2023 Signature of the Supervisor
Q. No.	1	2	3	4
MARKS	6	7	8	-
TOTAL 2 12 M				

(Signature)
9/11/23

Start Writing answer below this line.

$$\boxed{1\frac{3}{4} + \frac{1}{2} + \frac{1}{4} + 0 + 2} \quad \boxed{7} \quad \boxed{1\frac{4}{4} + 1\frac{4}{4}}$$

b a c ✓

Q.1)

a) → Capacity factor -

It is the relative migration of zone of solute represented by k'

Selectivity factor - It is the ratio of partition coefficient of more retained solute to the partition coefficient of less retained solute $\Rightarrow \alpha = \frac{k_B}{k_A}$

$\frac{3}{4}$

Retention time - Time required to elute half of the zone or 50% of the zone.

Dead time - The time t_M which first appears in the chromatogram



Dr. R. O. Ganjiwale
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

b) → HETP -

It is the height equivalent to theoretical plates.

It is defined as the length of column divided by the no. of theoretical plates.

- As the theoretical plates increases the height of column decreases in constant column length.

$$\text{HETP} = \frac{L}{N}$$

where, L = length of column

N = no. of theoretical plates

c) → Fermi's resonance -

It is the coupling between fundamental vibration ^{one} and overtone of some other vibration called Fermi's resonance.



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Mg/e), Wardha

Example of Fermi's resonance → $\text{HCHO} \rightleftharpoons \text{HCO} + \text{H}$

In aldehyde HCHO at 2nd overtone, coupling between C-H stretching and C=O stretching observed.

The C-H stretching vibrations of CHO group, interacts with the second harmonical level, of deformation vibration resulting of $-\text{CNO}$ group ($2 \times 1400 \text{ cm}^{-1}$) resulting into fermi doublet in the range of $1380 - 1280 \text{ cm}^{-1}$ in spectra.

d) \rightarrow the base of hydrogen coupling with two bands from HCHO due to C-H stretching and C=O stretching.



R. O. Ganjiwale
(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Mglic), Wardha

e) → 2D technique of development in TLC - Chromatography

- In this technique, the TLC plate used must be square shaped.
- The TLC plate is first placed in the chamber of first solvent system as normal TLC placed.
- Then after the first solvent system completes, the TLC plate is removed dried and then further applied for 2nd solvent system.
- In this, the TLC plate is rotated counter clockwise direction. Then plate is placed in 2nd solvent system and after that the development of plate occurs.
- This development used for the separation of mixture of components in TLC.



Dr. R. O. Ganjiwale
(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Q.2)

b) → Factors affecting vibrational frequency in IR Spectroscopy -

① Effect of Bond order -

- The bond order affects the position of absorption band in IR.
- The higher bond order increases the stretching frequency of absorption band.
- In C=C the Stretching frequency is higher than C-C and C-C Stretchers at 1200-800 cm⁻¹.
- In C≡N the Stretching frequency is higher than C=N and in C≡N stretching frequency is higher than that of C-N.

② Vibrational coupling -

- An isolated C-H shows only one type of stretching frequency.
- C-H bond in CH₂ group shows two types of stretching frequency one is symmetrical and another is antisymmetrical.
- In C-H bond of CH₃ group also symmetrical and antisymmetrical type of stretching frequency observed.
- C-H bond in CH₂ group and C-H bond in CH₃ group stretches at diffn frequency.



Dr. R. O. Ganjiwale
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

- Vibrational coupling occurs between two bonds vibrating with the same frequency, and they are reasonably close in the molecule.
- The vibrations may be both fundamental or fundamental vibrations with overtone of some other vibrations.

- In CO_2 group, two $\text{C}=\text{O}$ bonds with common carbon atom present.
- In these two types of fundamental stretching vibrations observed one is symmetrical and other antisymmetrical.
 - In symmetrical, the $\text{C}=\text{O}$ either stretches or contracts and absorption band shifts towards longer λ than that of carbonyl group. (IR inactive)
 - In antisymmetrical the $\text{C}=\text{O}$ stretches and contracts and absorption band shifts towards lower λ side than carbonyl group. (IR active)

③ Fermi's Resonance -

It is discovered by Enrico Fermi on the account of shifting of energies and intensities of absorption band in IR Spectroscopy.



Dr. R. O. Ganjiwale
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

- It is the coupling between fundamental vibration and orientation of some other vibration called Fermi's resonance.
 - In CHO group, C-H stretching vibration in CHO group interacts with second harmonic(energ) of deformation vibration of CHO group ($2 \times 1400 \text{ cm}^{-1}$) resulting in Fermi doublet observed at $1280 - 1080 \text{ cm}^{-1}$.
- (4) Hydrogen bonding -
- Hydrogen bonding affects the position and shape of absorption band.
 - There are two types of hydrogen bonding, Intermolecular and intramolecular.
 - In intramolecular H.B., occurs between the hydrogen bond of same molecule and lone pair of electron of same molecule.
 - In Intermolecular H.B., occurs between Hydrogen of one molecule and electronegative element of other molecule resulting in association.

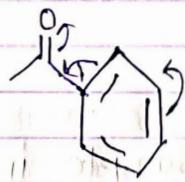


[Signature]
 (Dr. R. O. Ganjiwale)
 Principal
 PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

- The intermolecular H.B depends upon concentration of solution.
- The -OH stretching frequency in concentrated solution occurs at 3350 cm^{-1} and in dilute solution occurs at 3650 cm^{-1} .

⑤ Electronic effect

- Conjugation lowers the frequency in C=O and C=O stretching vibration.
- The delocalization of π electrons in C=O and in ring, increases the double bond character in ring.
- This results in decrease bond order which lowers the force constant and hence, lowering of stretching frequency by $20-30\text{ cm}^{-1}$.



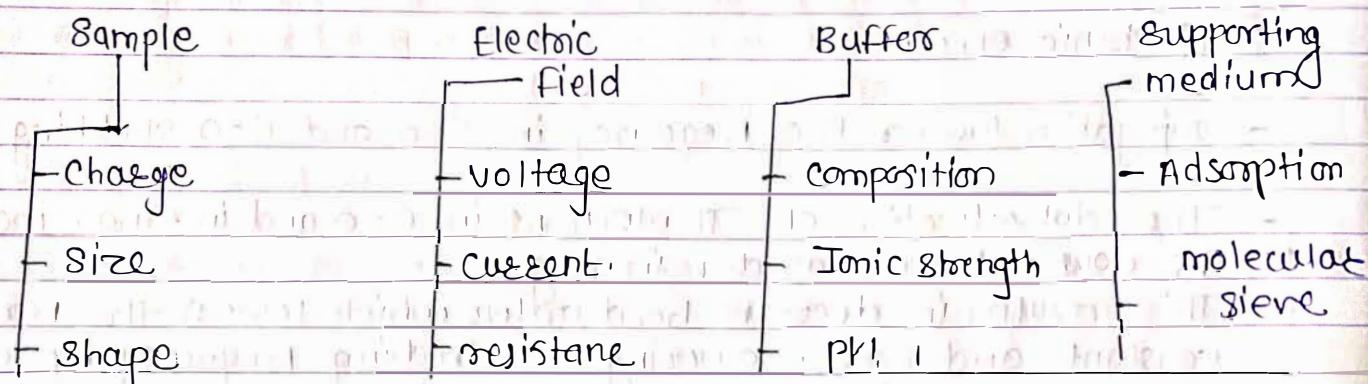
- This effect attributed because of mesomeric effect.



Dr. R. O. Ganjiwale
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Bargaon (Mglic), Wardha

Q.3)

c) → factors affecting electrophoretic mobility -



① Sample -

a) Charge - As the net charge of sample increases the rate of migration increases.

b) size - As the size of particle increases, the frictional and electrostatic force increases leading to decrease in rate of migration.



Dr. R. O. Ganjwale
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

- c) Shape - As shape ^{or particle} of migration state also changes.
- ② Electric field - By ohm's law,
- v = current \times voltage / resistance.
- a) Voltage - Increase in P.D or voltage increases the rate of migration of particles.
- b) Current - Increase in current increases rate of migration of particles.
- c) Resistance - Increase in resistance decreases rate of migration of particles.

③ Buffers -

- a) composition - acPlate buffers
potassium EDTA buffers
Phosphate buffers
- b) pH - pH increases ionization increases.



Dr. R. O. Ganjwale
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

c) Ionic strength - Ionic strength increases - The proportion of sample current through sample decreases and hence migration rate of particle decreases. As ionic strength decreases, the proportion of current passes through sample increases and hence increases the rate of migration of particles.

④ Supporting medium -

a) Adsorption - Increase in adsorption increases retention of particles in solution and hence rate of migration of particles decreases.



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

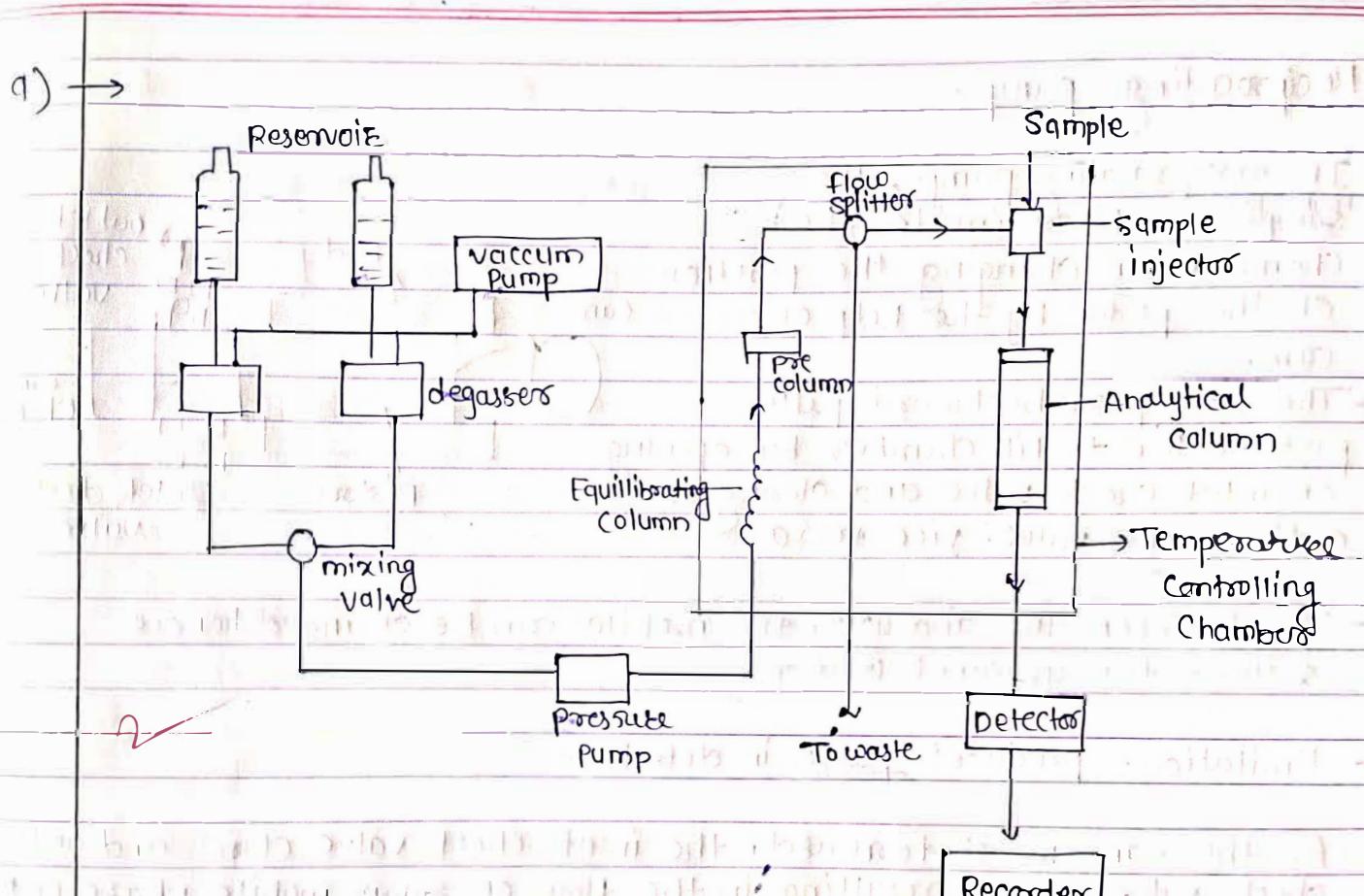


fig :- HPLC Unit



Dr. R. O. Ganjiwale
(Dr. R. O. Ganjiwale)
Principal

PRINCIPAL

Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Reciprocating pump -

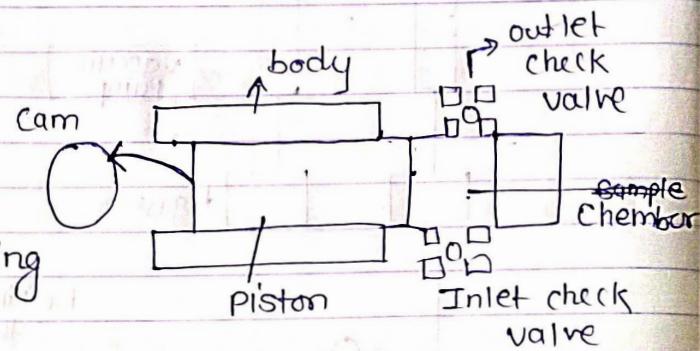
- In reciprocating pump, the mobile phase is filled inside the chamber by changing the position of the piston by the help of cam.

- The cam pushes backward pulls piston allow to fill chamber by opening of inlet check valve and closing of outlet check valve. vice versa.

- In between the analysis, the mobile can be changed hence suitable for gradient elution.

- Limitation - produces pulse in detector.

- As the cam moves forward, the inlet check valve closes and outlet check valve opens resulting in the flow of sample mobile phase outside the pump with high pressure.



(Dr. R. O. Ganjiwale)
Principal
PRINCIPALS
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Bulk property detector-

④ Refractive index detector-

- It works on change in the R.I of mobile phase. It detects the R.I of mobile phase.
- It is not suitable for gradient elution.
 - It is less sensitive than other detectors.
 - The changing of mobile phase in between analysis alters R.I and hence, interfere with detector response.



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Vidarbha Youth Welfare Society's
INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH
 WARDHA

~~Observe~~

PRACTICAL EXAMINATION

Second Sessional
 Practical Exam

Name Krushnali Sunil Thakare Examination Seat No. 15

Class B.P.M.-III^(Sem. II) Subject Instrumental Methods of Analysis Date 09/10/2023

Signature of the Supervisor

Q. No. 1 2 3 4 5 6 TOTAL

22
09/10/23

MARKS

Start Writing answer below this line.

Sl. No.	Contents	Maximum Marks	Marks Obtained
1)	Synopsis	10 M	6 -
2)	Major Experiment	25 M	19.5 -
3)	viva-voce	05 M	3.5 -
Total		40 M	

29
20 8

- Q.1) Synopsis
 a) Explain the principle of HPLC
 b) Draw Schematic diagram of GC.
 c) Differentiate between TLC and HPTLC
 d) What are some of the applications of Infrared Spectroscopy.
 e) Enlist most widely used carrier gases and detectors in Gas chromatography.



~~Observe~~
 (Dr. R. O. Ganjiwale)
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

a) → Principle of HPLC -

- In High performance liquid chromatography the separation is taken by between the two phases of mixture components on the basis of their affinity.
- After the sample injection through the rubber septum to the injector by syringe the required sample then passed through column.
- This column filled with liquid mobile phase which is elute through column by means of pressurised pump.
- This pressurised pump reduces the retention time of sample into column.
- Increases the efficiency by decreasing the time and increase reproducibility.

b) →

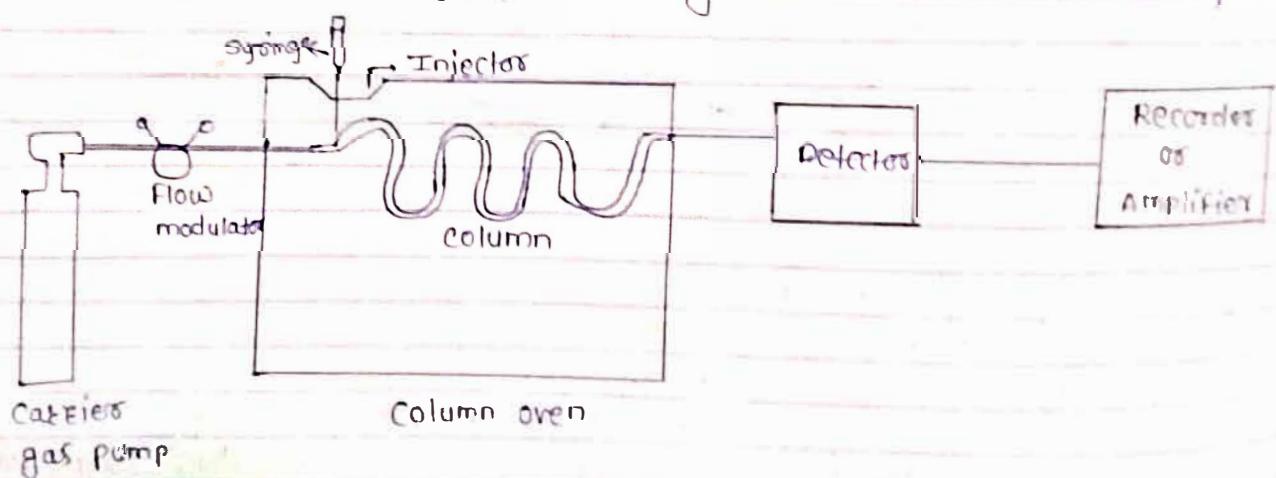


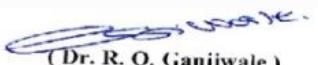
Fig:- Schematic diagram of GC



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

c) →	TLC	HPTLC
1)	<ul style="list-style-type: none"> - Thin layers chromatography, consist of TLC plate which run under the influence of capillary action. - In TLC, silica or activated charcoal used as a stationary phase on glass or aluminium plate. - It provides less precise results 	<ul style="list-style-type: none"> - In HPTLC, there is use of column enclosed in Sophisticated manner connected with Software programming. - In HPTLC, various size graded high efficiency columns are utilized. - It provides more precise results
d) →	<p>IR used in <u>Radar</u> region.</p> <p>IR spectroscopy used in identification of various components.</p> 	




 (Dr. R. O. Ganjiwale)
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Mehic), Wardha

e) → carrier gases

- Hydrogen, Helium, lithium, Argon, Nitrogen

Detectors

- Thermal conductivity detector
- Thermionic detector
- flame Ionization detector
- Electron capture detector

81



(Dr. R. O. Ganjiwale)
Principal

PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

Vidarbha Youth Welfare Society's
INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

WARDHA

Instrumental methods of
Analysis

Examination Seat No. 15

Subject

Signature of the Supervisor

Supplement No.

Date 09/10/2023

Start Writing answer below this line.

Q.2) Major Experiment

Aim :- To perform assay of paracetamol tablet as per IP by reference standard method.

Observation :-

1] Marketed paracetamol tablet

a) weight of tablet = 11.94 gm

b) Average wt. of tablet = 0.597 gm

c) Label claim = 0.500 gm

d) 0.597 gm of tablet powder paracetamol = 0.500 gm of paracetamol

$$0.597 \text{ gm} \longrightarrow 0.500 \text{ gm of paracetamol}$$

$$\underline{x \text{ gm}} \longrightarrow 0.150 \text{ gm of paracetamol}$$

$$x = \frac{0.597 \times 0.150}{0.500}$$

$$x = \underline{\underline{0.179 \text{ gm}}}$$



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghc), Wardha

Absorbance of Reference = 0.325
 Absorbance of Sample = 0.281

Calculation of Tablet Content -

$$\begin{aligned} \text{Tablet Content} &= \frac{A_{257 \text{ nm}}(\text{Sample})}{A_{257 \text{ nm}}(\text{Reference})} \times \frac{\text{Reference Sample wt. of Sample taken}}{\text{wt. of Sample taken}} \times \frac{\text{Avg. wt. of tablet}}{\text{Label claim}} \\ &= \frac{0.281}{0.325} \times \frac{0.150}{0.179} \times \frac{0.597}{0.500} \\ &= 0.86 \times 0.8379 \times 1.194 \\ &= \underline{\underline{0.859 \text{ gm}}} \end{aligned}$$

$$\begin{aligned} \% \text{ Paracetamol} &= \frac{A_{257 \text{ nm}}(\text{Sample})}{A_{257 \text{ nm}}(\text{Reference})} \times \frac{\text{wt. of Reference sample}}{\text{wt. of Sample taken}} \times \frac{\text{Avg. wt. of tablet}}{\text{Label claim}} \times 100 \\ &= \frac{0.281}{0.325} \times \frac{0.150}{0.179} \times \frac{0.597}{0.500} \times 100 \end{aligned}$$



(Dr. R. O. Ganjiwale)
 Principal
 PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

$$\begin{aligned}
 &= 0.859 \times 0.8379 \times 1.194 \times 100 \\
 &= \underline{\underline{85.84 \%}}
 \end{aligned}$$

Result :-

- 1) The content of paracetamol of given paracetamol tablet was found to be 0.859 gm
- 2) The % purity of given sample of 500mg of paracetamol tablet as per IP was found to be 85.84 %

(a)



Dr. R. O. Ganjiwale
 Principal
 PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Megh), Wardha

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

First / Second Sessional Attendance Sheet 2023 - 2024
B.PHARM - IV (SEM-VII)

Theory / Practical

Subject : Instrumental Methods of Analysis

Date : 04/11/2023

Sr. No.	Name of Students	Signature
BATCH: 'A'		
1	Ku. A. M. Kolhe	<u>Kolhe</u>
2	Ku. A. N. Shaikh	<u>Shaikh</u>
3	Ku. A. P. Gote	<u>Gote</u>
4	Ku. A. S. Talware	<u>Talware</u>
5	Ku. A. S. Tinghase	<u>Tinghase</u>
6	Ku. B. P. Thakare	<u>Thakare</u>
7	Ku. C. R. Shende	<u>Shende</u>
8	Ku. D. N. Mandaokar	<u>Mandaokar</u>
9	Ku. E. U. Kotharkar	<u>Kotharkar</u>
10	Ku. G. S. Kawle	<u>Kawle</u>
11	Ku. I. S. Borkar	<u>Borkar</u>
12	Ku. J. S. Satone	<u>Satone</u>
13	Ku. K. D. Bele	<u>(K) Bele</u>
14	Ku. K. R. Raut	<u>(K) Raut</u>
15	Ku. K. S. Thakare	<u>Thakare</u>
16	Ku. L. S. Mahajan	<u>Mahajan</u>
17	Ku. M. G. Sayam	<u>Sayam</u>
18	Ku. M. N. Hatwar	<u>Hatwar</u>
19	Ku. M. R. Verma	<u>M. R. Verma</u>
20	Ku. P. R. Umate	<u>Umate</u>
21	Ku. P. V. Lanjekar	<u>P. V. Lanjekar</u>
22	Ku. R. K. Umathé	<u>Umathé</u>
23	Ku. R. M. Shidodkar	<u>Shidodkar</u>
24	Ku. R. P. Deshpande	<u>Deshpande</u>
25	Ku. R. S. Khurpade	<u>Khurpade</u>
BATCH: 'B'		
26	Ku. R. S. Raut	<u>Raut</u>
27	Ku. R. V. Bhale	<u>Bhale</u>
28	Ku. S. M. Yadav	<u>Yadav</u>
29	Ku. S. N. Deshmukh	<u>Deshmukh</u>
30	Ku. S. P. Tapre	<u>Tapre</u>
31	Ku. S. R. Chahande	<u>Chahande</u>
32	Ku. S. R. Thakare	<u>Thakare</u>
33	Ku. S. R. Tiwari	<u>Tiwari</u>
34	Ku. S. S. Ahirrao	<u>Ahirrao</u>

Room No. 311

Present Student - 36
Absent Student - 04
Total - 40

{
Attendance
04.11.23}



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

35	Ku.	S. S.	Misalkar	<i>✓✓✓</i>
36	Ku.	S. S.	Nimbalkar	<i>✓✓✓</i>
37	Ku.	S. T.	Yedlawar	<i>✓✓✓</i>
38	Ku.	S. V.	Bhoyar	<i>✓✓✓</i>
39	Ku.	S. V.	Dhok	<i>✓✓✓</i>
40	Ku.	Sakshi N.	Raut	<i>✓✓✓</i>
41	Ku.	Shreya S.	Raut	<i>✓✓✓</i>
42	Ku.	T. R.	Dhage	<i>T. R. D. H. G.</i>
43	Ku.	T. R.	Hukum	<i>Hukum</i>
44	Ku.	T. U.	Moon	<i>Moon</i>
45	Ku.	U. S.	Gawande	<i>Gawande</i>
46	Ku.	V. A.	Kshirsagar	<i>Kshirsagar</i>
47	Ku.	V. D.	More	<i>More</i>
48	Ku.	V. D.	Shrivastava	<i>Shrivastava</i>
49	Ku.	V. N.	Deshmukh	<i>Deshmukh</i>
50	Ku.	V. R.	Sushir	<i>Sushir</i>
BATCH: 'C'				
51	Ku.	Y. D.	Kude	<i>Kude</i>
52	Mr.	A. C.	Darda	<i>Darda</i>
53	Mr.	A. K.	Kinhekar	<i>Kinhekar</i>
54	Mr.	A. N.	Dahake	<i>Dahake</i>
55	Mr.	G. M.	Sukhadia	<i>Sukhadia</i>
56	Mr.	H. C.	Butale	<i>Butale</i>
57	Mr.	J. S.	Kadu	<i>Kadu</i>
58	Mr.	M. A.	Hydri	<i>Hydri</i>
59	Mr.	N. O.	Dhanbhate	<i>Dhanbhate</i>
60	Mr.	N. S.	Chhajed	<i>Chhajed</i>
61	Mr.	O. P.	Patte	<i>Patte</i>
62	Mr.	P. M.	Zade	<i>Zade</i>
63	Mr.	P. S.	Waghe	<i>Waghe</i>
64	Mr.	P. S.	Wanare	<i>Wanare</i>
65	Mr.	R. G.	Bongade	<i>Bongade</i>
66	Mr.	R. G.	Prabhakar	<i>Prabhakar</i>
67	Mr.	R. R.	Magar	<i>Magar</i>
68	Mr.	R. S.	Dhumale	<i>Dhumale</i>
69	Mr.	R. S.	Jawade	<i>Jawade</i>
70	Mr.	S. B.	Fulmalgi	<i>Fulmalgi</i>
71	Mr.	S. C.	Borkar	<i>Borkar</i>
72	Mr.	S. D.	Dahare	<i>Dahare</i>
73	Mr.	S. H.	Gawali	<i>Gawali</i>
74	Mr.	S. R.	Bokade	<i>Bokade</i>
75	Mr.	T. M.	Khade	<i>Khade</i>
76	Mr.	Y. M.	Karlekar	<i>Karlekar</i>
77	Mr.	Y. R.	Kondawar	<i>Kondawar</i>

Room NO - 313

Present = 35

Absent = 02

(Dr. R. O. Ganjwale)
Principal

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

First / Second Sessional Attendance Sheet 2023 - 2024
B.PHARM - IV (SEM-VII)

Theory / Practical

Subject : Instrumental method of analysis
Practical.

Date : 9 - 10 - 23.

Sr. No.	Name of Students			Signature
BATCH: 'A'				
1	Ku.	A. M.	Kolhe	<i>SAP Kolhe</i>
2	Ku.	A. N.	Shaikh	<i>Ashaikh</i>
3	Ku.	A. P.	Gote	<i>Afote</i>
4	Ku.	A. S.	Talware	<i>A. Talware</i>
5	Ku.	A. S.	Tinghase	<i>Angata</i>
6	Ku.	B. P.	Thakare	<i>B. Thakare</i>
7	Ku.	C. R.	Shende	<i>Chunchu</i>
8	Ku.	D. N.	Mandaokar	<i>D. Mandaokar</i>
9	Ku.	E. U.	Kotharkar	<i>Kotharkar</i>
10	Ku.	G. S.	Kawle	<i>AB</i>
11	Ku.	I. S.	Borkar	<i>E. Borkar</i>
12	Ku.	J. S.	Satone	<i>AB</i>
13	Ku.	K. D.	Bele	<i>R. Bele</i>
14	Ku.	K. R.	Raut	<i>R.R. Raut</i>
15	Ku.	K. S.	Thakare	<i>K. Thakare</i>
16	Ku.	L. S.	Mahajan	<i>L. Mahajan</i>
17	Ku.	M. G.	Sayam	<i>M. Sayam</i>
18	Ku.	M. N.	Hatwar	<i>M. Hatwar</i>
19	Ku.	M. R.	Verma	<i>M. Verma</i>
20	Ku.	P. R.	Umate	<i>P. Umate</i>
21	Ku.	P. V.	Lanjekar	<i>P.V. Lanjekar</i>
22	Ku.	R. K.	Umathé	<i>R. Umathé</i>
23	Ku.	R. M.	Shidodkar	<i>R. Shidodkar</i>
24	Ku.	R. P.	Deshpande	<i>AB</i>
25	Ku.	R. S.	Khurpade	<i>R. Khurpade</i>
BATCH: 'B'				
26	Ku.	R. S.	Raut	<i>Raut</i>
27	Ku.	R. V.	Bhale	<i>R. Bhale</i>
28	Ku.	S. M.	Yadav	<i>S. Yadav</i>
29	Ku.	S. N.	Deshmukh	<i>S. Deshmukh</i>
30	Ku.	S. P.	Tapre	<i>S. Tapre</i>
31	Ku.	S. R.	Chahande	<i>S. Chahande</i>
32	Ku.	S. R.	Thakare	<i>S. Thakare</i>
33	Ku.	S. R.	Tiwari	<i>S. Tiwari</i>
34	Ku.	S. S.	Ahirrao	<i>S. Ahirrao</i>

Total No. of Students = 71
Present = 72
Absent = 05

9/10/23

35	Ku.	S. S.	Misalkar	<i>Bijin</i>
36	Ku.	S. S.	Nimbalkar	<i>(S. S.)</i>
37	Ku.	S. T.	Yedlawar	<i>(S. T.)</i>
38	Ku.	S. V.	Bhoyar	<i>(S. V.)</i>
39	Ku.	S. V.	Dhok	<i>(Dhok)</i>
40	Ku.	Sakshi N.	Raut	<i>(Raut)</i>
41	Ku.	Shreya S.	Raut	<i>(Shreya S.)</i>
42	Ku.	T. R.	Dhage	<i>(T. R.)</i>
43	Ku.	T. R.	Hukum	<i>(Hukum)</i>
44	Ku.	T. U.	Moon	<i>(Moon)</i>
45	Ku.	U. S.	Gawande	<i>AB</i>
46	Ku.	V. A.	Kshirsagar	<i>(Kshirsagar)</i>
47	Ku.	V. D.	More	<i>(More)</i>
48	Ku.	V. D.	Shrivastava	<i>(Shrivastava)</i>
49	Ku.	V. N.	Deshmukh	<i>(Deshmukh)</i>
50	Ku.	V. R.	Sushir	<i>(Sushir)</i>
BATCH 'C'				
51	Ku.	Y. D.	Kude	<i>(Kude)</i>
52	Mr.	A. C.	Darda	<i>(Darda)</i>
53	Mr.	A. K.	Kinhekar	<i>(Kinhekar)</i>
54	Mr.	A. N.	Dahake	<i>(Dahake A. N.)</i>
55	Mr.	G. M.	Sukhadia	<i>(Sukhadia)</i>
56	Mr.	H. C.	Butale	<i>(Butale)</i>
57	Mr.	J. S.	Kadu	<i>(Kadu)</i>
58	Mr.	M. A.	Hydri	<i>(Hydri)</i>
59	Mr.	N. O.	Dhanbhate	<i>(Dhanbhate)</i>
60	Mr.	N. S.	Chhajed	<i>(Chhajed)</i>
61	Mr.	O. P.	Patte	<i>(Patte)</i>
62	Mr.	P. M.	Zade	<i>(Zade)</i>
63	Mr.	P. S.	Waghe	<i>(Waghe)</i>
64	Mr.	P. S.	Wanare	<i>(Wanare)</i>
65	Mr.	R. G.	Bongade	<i>(Bongade)</i>
66	Mr.	R. G.	Prabhakar	<i>(Prabhakar)</i>
67	Mr.	R. R.	Magar	<i>(Magar)</i>
68	Mr.	R. S.	Dhumale	<i>AB</i>
69	Mr.	R. S.	Jawade	<i>(Jawade)</i>
70	Mr.	S. B.	Fulmalgi	<i>(Fulmalgi)</i>
71	Mr.	S. C.	Borkar	<i>(Borkar)</i>
72	Mr.	S. D.	Dahare	<i>(Dahare)</i>
73	Mr.	S. H.	Gawali	<i>(Gawali)</i>
74	Mr.	S. R.	Bokade	<i>(Bokade)</i>
75	Mr.	T. M.	Khade	<i>(Khade)</i>
76	Mr.	Y. M.	Karlekar	<i>(Karlekar)</i>
77	Mr.	Y. R.	Kondawar	<i>(Kondawar)</i>



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Meghe), Wardha

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

III Sessional Examination : 2023 - 24

D.Pharm ____ B.Pharm IV (Sem. VII) M.Pharm ____ (Sem. ____)

RESULT SHEET

Subject : Instrumental Methods of analysis Max. Marks 30

Exam. Seat No.	Marks Obtained question-wise						Total Marks 30
	1(10)	2(10)	3(10)	4	5	6	
1	2½	8½	4½				15
2	8½	7	7				22½
3	2	7	7½				16½
4	7½	6	6				12½
5	1½	6½	7½				15½
6	0	7	3¾				10¾
7	3¾	7½	6¾				18
8	A	B	S	E	N	T	-
9	5¾	7½	7½				20¾
10	A	B	S	E	N	T	-
11	3	6½	2				11½
12	A	B	S	E	N	T	-
13	4	7½	8½				20
14	5	7½	7½				20
15	6½	7	8				21½
16	3	7	6				16
17	2½	7	7½				17
18	2¾	7	3½				13¼
19	2¾	7	7				16¾
20	5½	6	7½				19
21	0	4½	3				7½
22	2½	7	4				13½
23	5¾	4	7½				17¼
24	A	B	S	E	N	T	-
25	2¾	6	4				12¾
26	1¼	7½	6½				15
27	2½	7	3				12½
28	4¾	7	7½				19
29	3½	7	7				17½
30	½	4	0				4½
31	3	6½	5¾				15¼

Absent Number 8, 10, 12, 24

Signature of Examiner : M. S. WILSON

Date 6. XI. 23

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

I/I/Sessional Examination : 2023-24

D.Pharm ____ B.Pharm ____ (Sem. VII) M.Pharm ____ (Sem. ____)

RESULT SHEET

Subject : Instrumental Methods of Analysis Max. Marks 30

Exam. Seat No.	Marks Obtained question-wise						Total Marks 30
	1(10)	2(10)	3(10)	4	5	6	
32	3 3/4 ✓	5 1/4 ✓	7 1/2 ✓				16 1/2 ✓
33	1 1/2 ✓	6 ✓	4 1/2 ✓				12 ✓
34	3 3/4 ✓	6 ✓	7 ✓				16 3/4 ✓
35	6 ✓	6 1/2 ✓	7 ✓				19 1/2 ✓
36	2 ✓	6 1/2 ✓	3 1/2 ✓				12 ✓
37	4 3/4 ✓	7 ✓	7 1/2 ✓				19 1/4 ✓
38	0 ✓	3 ✓	3 1/2 ✓				6 1/2 ✓
39	5 1/2 ✓	6 ✓	6 1/2 ✓				18 ✓
40	2 1/4 ✓	7 ✓	5 3/4 ✓				15 ✓
41	1 ✓	5 1/2 ✓	5 1/2 ✓				12 ✓
42	8 1/2 ✓	5 ✓	8 ✓				21 1/2 ✓
43	4 ✓	8 ✓	7 1/2 ✓				19 1/2 ✓
44	3 ✓	6 ✓	8 ✓				17 ✓
45	A	B	C	E	N	T	-
46	2 1/2 ✓	8 ✓	7 1/2 ✓				18 ✓
47	6 1/2 ✓	7 ✓	8 ✓				21 1/2 ✓
48	6 ✓	7 ✓	8 ✓				21 ✓
49	7 3/4 ✓	7 1/2 ✓	8 ✓				23 1/4 ✓
50	3 ✓	6 ✓	8 ✓				17 ✓
51	6 1/2 ✓	8 1/4 ✓	8 ✓				22 3/4 ✓
52	1/4 ✓	1 1/2 ✓	4 ✓				5 3/4 ✓
53	1 ✓	4 ✓	4 ✓				9 ✓
54	2 ✓	7 ✓	3 1/2 ✓				12 1/2 ✓
55	1 ✓	8 ✓	8 ✓				17 ✓
56	3 3/4 ✓	7 1/2 ✓	4 ✓				15 1/4 ✓
57	4 ✓	1 ✓	5 ✓				10 ✓
58	1 3/4 ✓	7 1/2 ✓	4 ✓				13 1/4 ✓
59	1 1/4 ✓	6 3/4 ✓	7 ✓				15 ✓
60	1 ✓	3 ✓	4 1/2 ✓				8 1/2 ✓
61	1 ✓	7 1/2 ✓	4 ✓				12 1/2 ✓
62	1 3/4 ✓	7 ✓	6 1/4 ✓				15 ✓

Absent Number 45

Signature of Examiner : P.W. Patil

Date 6.11.23

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH**Borgaon (Meghe), Wardha**

I/I Sessional Examination : 2023-24

D.Pharm ____ B.Pharm IV (Sem. VII) M.Pharm ____ (Sem. ____)**RESULT SHEET**Subject : Instrumental Methods of Analysis Max. Marks 30

Exam. Seat No.	Marks Obtained question-wise						Total Marks
	1(10)	2(10)	3(10)	4	5	6	
63	4 ✓	6 1/2 ✓	6 ✓				<u>16 1/2</u>
64	6 1/4 ✓	7 ✓	7 ✓				<u>20 1/4</u>
65	2 1/4 ✓	7 ✓	7 1/2 ✓				<u>16 3/4</u>
66	3 1/4 ✓	6 3/4 ✓	7 ✓				<u>17</u>
67	1 3/4 ✓	0 ✓	2 ✓				<u>3 3/4</u>
68	A	B	S	E	N	T	—
69	1 3/4 ✓	7 ✓	4 ✓				<u>12 3/4</u>
70	3/4 ✓	7 ✓	3 1/4 ✓				<u>11</u>
71	1 3/4 ✓	6 3/4 ✓	6 1/2 ✓				<u>15</u>
72	7 3/4 ✓	6 1/2 ✓	7 1/4 ✓				<u>21 1/2</u>
73	1/2 ✓	5 ✓	4 ✓				<u>9 1/2</u>
74	0 ✓	0 ✓	2 1/2 ✓				<u>2 1/2</u>
75	1 ✓	7 ✓	7 1/2 ✓				<u>15 1/2</u>
76	5 ✓	6 ✓	4 ✓				<u>15</u>
77	2 ✓	7 ✓	4 ✓				<u>13</u>
.....
.....
.....
.....
.....
.....
.....
.....
.....
.....
.....
.....
.....
.....
.....
.....

Absent Number 6, 8Signature of Examiner : R.M. Patil Date 6. X. 23 36

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

I/I/Sessional Examination : Practical - 23-24
D.Pharm ____ B.Pharm IV (Sem. VII) M.Pharm ____ (Sem. ____)

RESULT SHEET

Subject : Instrumental method of Analysis Max. Marks 40 M

Exam. Seat No.	Marks Obtained question-wise						Total Marks <u>40M</u>
	<u>Q-1=10M</u>	<u>Q-2=25M</u>	<u>Q-3=5</u>	4	5	6	
1	<u>3.5</u>	<u>20</u>	<u>3.5</u>				<u>27</u>
2	<u>5.5</u>	<u>20.5</u>	<u>04</u>				<u>30</u>
3	<u>5.5</u>	<u>19</u>	<u>03</u>				<u>27.5</u>
4	<u>0.5</u>	<u>19</u>	<u>3.5</u>				<u>23</u>
5	<u>04</u>	<u>19</u>	<u>03</u>				<u>26</u>
6	<u>2.5</u>	<u>20</u>	<u>3.5</u>				<u>26</u>
7	<u>4.5</u>	<u>20</u>	<u>3.5</u>				<u>28</u>
8	<u>2.5</u>	<u>18</u>	<u>2.5</u>				<u>23</u>
9	<u>05</u>	<u>20</u>	<u>03</u>				<u>28</u>
10	<u>AB</u>	<u>AB</u>	—	—	—	—	<u>AB</u>
11	<u>04</u>	<u>21</u>	<u>03</u>				<u>28</u>
12	<u>AB</u>	<u>AB</u>	—	—	—	—	<u>AB</u>
13	<u>02</u>	<u>21</u>	<u>03</u>				<u>26</u>
14	<u>07</u>	<u>22.5</u>	<u>3.5</u>				<u>33</u>
15	<u>06</u>	<u>19.5</u>	<u>3.5</u>				<u>29</u>
16	<u>5.5</u>	<u>19.5</u>	<u>03</u>				<u>28</u>
17	<u>06</u>	<u>19</u>	<u>03</u>				<u>28</u>
18	<u>05</u>	<u>18.5</u>	<u>3.5</u>				<u>27</u>
19	<u>03</u>	<u>20</u>	<u>03</u>				<u>28</u>
20	<u>05</u>	<u>18.5</u>	<u>3.5</u>				<u>27</u>
21	<u>1.5</u>	<u>20</u>	<u>2.5</u>				<u>24</u>
22	<u>2.5</u>	<u>20</u>	<u>3.5</u>				<u>28</u>
23	<u>05</u>	<u>20</u>	<u>04</u>				<u>29</u>
24	<u>AB</u>	<u>AB</u>	—	—	—	—	<u>AB</u>
25	<u>04</u>	<u>20</u>	<u>03</u>				<u>27</u>

Absent Number 03

Signature of Examiner :

Date 12/10/23

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

D.Pharm _____ B.Pharm (Sem. VII) M.Pharm _____
Sessional Examination : Baetra 23-24**RESULT SHEET**Subject : Instrumental method of Analysis Max. Marks 40M

Exam. Seat No.	Marks Obtained question-wise						Total Marks
	Q=1-10M	Q=2=25M	Q=3=5M	4	5	6	
26	2.5	21	3.5				40M
27	5.5	21.5	03				27
28	2.5	21	3.5				30
29	3.5	21.5	04				27
30	01	22	02				29
31	05	21	03				25
32	5.5	20	03				29
33	06	20	04				30
34	3.5	22	3.5				29
35	05	21	03				29
36	02	21	2.5				25.5
37	3.5	20	2.5				26
38	07	20	02				22
39	05	20	03				28
40	03	22.5	3.5				29
41	03	21	03				27
42	6.5	20	3.5				28
43	04	21	02				27
44	05	20	03				28
45	AB	AB	-	-	-		AB
46	03	21	03				27
47	6.5	22.5	04				33
48	06	21	04				31
49	6.5	22	04				32.5
50	02	21	03				26

Absent Number 01

Signature of Examiner : J. D. J. Date 12/10/23

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Borgaon (Meghe), Wardha

D.Pharm B.Pharm IV (Sem. VII) M.Pharm (Sem.)

RESULT SHEET

Subject : Instrumental method of Analysis Max. Marks 10M

Exam. Seat No.	Marks Obtained question-wise						Total Marks
	Q-1-10M	Q-2-25M	Q-3-5M	4	5	6	
51	6.5	19.5	03				40M
52	02	20	02				29
53	4.5	20	2.5				24
54	04	19	03				27
55	07	20	04				26
56	05	20	03				31
57	1.5	20	1.5				28
58	03	20	02				23
59	1.5	19.5	02				25
60	3.5	20	2.5				23
61	02	19	02				26
62	3.5	20	2.5				29
63	6.5	20	2.5				25
64	04	19	02				27
65	06	20	03				29
66	07	20	03				30
67	1.5	20	1.5				23
68	AB	AB	-	-	-	-	413
69	01	20	01	-	-	-	22
70	03	19	03				25
71	01	18	01				20
72	04	20	03				27
73	01	18	01				20
74	02	17	01				20
75	3.5	20	2.5				26
76	04	20	03				27
77	3.5	18.5	02				24

Absent Number 01

Signature of Examiner : 

Date 12/10/23

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH
 Borgaon (Meghe) Wardha

SECOND SESSIONAL EXAMINATION, 2023-2024
 B.PHARM - IV (SEM - VII)
 *** RESULT - SHEET ***

Sr. No.	Name of Students	Instrumental Methods of Analysis	Industrial Pharmacy II	Pharmacy Practice	Novel Drug Delivery System	Total	Result
1	Ku. A. M. Kolhe	15	20.5	23	19	77.5	PASS
2	Ku. A. N. Shaikh	22.5	24	24	21	91.5	PASS
3	Ku. A. P. Gote	16.5	20	20	18	74.5	PASS
4	Ku. A. S. Talware	12.5	17.5	20	17	67	FAIL
5	Ku. A. S. Tinghase	15.5	11	20	17.5	64	FAIL
6	Ku. B. P. Thakare	10.75	18	21	12.5	62.25	FAIL
7	Ku. C. R. Shende	18	21	21	19	79	PASS
8	Ku. D. N. Mandaokar	AB	16	20	13.5	49.5	FAIL
9	Ku. E. U. Kotharkar	20.75	20	21	20	81.75	PASS
10	Ku. G. S. Kawle	AB	AB	0	AB	0	FAIL
11	Ku. I. S. Borkar	11.5	21	21	19	72.5	FAIL
12	Ku. J. S. Satone	AB	AB	0	AB	0	FAIL
13	Ku. K. D. Bele	20	22.5	24	20	86.5	PASS
14	Ku. K. R. Raut	20	24.5	26	22	92.5	PASS
15	Ku. K. S. Thakare	21.5	26.5	26	25	99	PASS
16	Ku. L. S. Mahajan	16	23	24	24.5	87.5	PASS
17	Ku. M. G. Sayam	17	19.5	24	24.5	85	PASS
18	Ku. M. N. Hatwar	13.25	20.5	22	19	74.75	FAIL
19	Ku. M. R. Verma	16.75	22.5	25	21.5	85.75	PASS
20	Ku. P. R. Umate	19	26	22	19	86	PASS
21	Ku. P. V. Lanjekar	7.25	10	16	13.5	46.75	FAIL
22	Ku. R. K. Umathe	13.5	17.5	21	21	73	FAIL
23	Ku. R. M. Shidodkar	17.25	22.5	24	23	86.75	PASS
24	Ku. R. P. Deshpande	AB	AB	0	AB	0	FAIL
25	Ku. R. S. Khurpade	12.75	18	20	18.5	69.25	FAIL
26	Ku. R. S. Raut	15	16.5	22	17.5	71	PASS
27	Ku. R. V. Bhale	12.25	24	21	23	80.25	FAIL
28	Ku. S. M. Yadav	19	24.5	26	24.5	94	PASS
29	Ku. S. N. Deshmukh	17.5	21	25	22	85.5	PASS
30	Ku. S. P. Tapre	4.5	13	15	10.5	43	FAIL




 Dr. R. O. Ganjiwale
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

31	Ku.	S. R.	Chahande	15.25	17.5	21	22.5	76.25	PASS
32	Ku.	S. R.	Thakare	16.5	17	24	18.5	76	PASS
33	Ku.	S. R.	Tiwari	12	19	23	19	73	FAIL
34	Ku.	S. S.	Ahirrao	16.75	19	21	22	78.75	PASS
35	Ku.	S. S.	Misalkar	19.5	23.5	22	23.5	88.5	PASS
36	Ku.	S. S.	Nimbalkar	12	17	19	22	70	FAIL
37	Ku.	S. T.	Yedlawar	19.25	21	22	21	83.25	PASS
38	Ku.	S. V.	Bhoyar	6.5	12.5	19	7.5	45.5	FAIL
39	Ku.	S. V.	Dhok	18	24	25	23	90	PASS
40	Ku.	Sakshi	Raut	15	21	18	17	71	PASS
41	Ku.	Shreya	S Raut	12	16	18	17.5	63.5	FAIL
42	Ku.	T. R.	Dhage	21.5	20.5	20	20.5	82.5	PASS
43	Ku.	T. R.	Hukum	19.5	23	23	25	90.5	PASS
44	Ku.	T. U.	Moon	17	15	22	21	75	PASS
45	Ku.	U. S.	Gawande	AB	AB	0	AB	0	FAIL
46	Ku.	V. A.	Kshirsagar	18	18.5	21	18.5	76	PASS
47	Ku.	V. D.	More	21.5	20	23	25	89.5	PASS
48	Ku.	V. D.	Shrivastava	21	23	23	28	95	PASS
49	Ku.	V. N.	Deshmukh	23.25	21	26	20.5	90.75	PASS
50	Ku.	V. R.	Sushir	17	24	22	23	86	PASS
51	Ku.	Y. D.	Kude	22.75	23	20	25.5	91.25	PASS
52	Mr.	A. C.	Darda	5.75	9.5	16	14.5	45.75	FAIL
53	Mr.	A. K.	Kinhekar	9	12.5	14	15.5	51	FAIL
54	Mr.	A. N.	Dahake	12.5	12.5	18	13	56	FAIL
55	Mr.	G. M.	Sukhadia	17	21	23	21.5	82.5	PASS
56	Mr.	H. C.	Butale	15.25	15	14	19	63.25	FAIL
57	Mr.	J. S.	Kadu	10	4	18	11.5	43.5	FAIL
58	Mr.	M. A.	Hydri	13.25	19	21	16	69.25	FAIL
59	Mr.	N. O.	Dhanbhate	15	16.5	22	13.5	67	FAIL
60	Mr.	N. S.	Chhajed	8.5	16	18	20	62.5	FAIL
61	Mr.	O. P.	Patte	12.5	13	14	10.5	50	FAIL
62	Mr.	P. M.	Zade	15	18.5	23	21.5	78	PASS
63	Mr.	P. S.	Waghe	16.5	20	22	19	77.5	PASS
64	Mr.	P. S.	Wanare	20.25	25.5	18	20.5	84.25	PASS
65	Mr.	R. G.	Bongade	16.75	26	23	24	89.75	PASS
66	Mr.	R. G.	Prabhakar	17	26.5	19	21	83.5	PASS
67	Mr.	R. R.	Magar	3.75	7.5	12	12	35.25	FAIL
68	Mr.	R. S.	Dhumale	AB	AB	0	AB	0	FAIL



(Dr. R. O. Ganjiwale)
 Principal
 PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghc), Wardha

69	Mr. R. S. Jawade	12.75	8	17	16	53.75	FAIL
70	Mr. S. B. Fulmali	11	15	16	11.5	53.5	FAIL
71	Mr. S. C. Borkar	15	17	19	18.5	69.5	PASS
72	Mr. S. D. Dahare	21.5	25.5	20	26	93	PASS
73	Mr. S. H. Gawali	9.5	12.5	15	15	52	FAIL
74	Mr. S. R. Bokade	2.5	3	10	9	24.5	FAIL
75	Mr. T. M. Khade	15.5	20	19	18	72.5	PASS
76	Mr. Y. M. Karlekar	15	20.5	17	15	67.5	PASS
77	Mr. Y. R. Kondawar	13	12.5	17	16.5	59	FAIL
Appeared:		71	72	77	72	77	
Passed:		45	58	67	59	42	
Result in Percentage:		63.38%	80.56%	87.01%	81.94%	54.55%	

Date 09/11/2023

Entered By:-
(Mr. V. A. Rekhate)

Checked By:-
Ms. S. P. Gautam & Dr. B. R. Gandhare

(Padhav)

(Signature)
(Dr. R. O. Ganjiwale)
Principal
. PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Megh), Wardha



(Signature)
(Dr. R. O. Ganjiwale)
Principal
. PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Megh), Wardha

INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH
 Borgaon (Meghe) Wardha

INTERNAL MARKS (Average + Continuous Mode) , 2023 - 2024
B.PHARM-IV (Sem-VII)

S.N.	Name of Candidate	Average Marks Theory				Average Marks Practical	Total
		Instrumental Methods of Analysis	Industrial Pharmacy	Pharmacy Practice	Novel Drug Delivery System		
1	Ku. A. M. Kolhe	15	18	16	16	9	74.00
2	Ku. A. N. Shaikh	16	21	20	18	12	87.00
3	Ku. A. P. Gote	17	19	19	17	11	83.00
4	Ku. A. S. Talware	16	14	16	16	9	71.00
5	Ku. A. S. Tinghase	16	14	20	16	11	77.00
6	Ku. B. P. Thakare	10	12	17	13	10	62.00
7	Ku. C. R. Shende	16	17	16	17	12	78.00
8	Ku. D. N. Mandaokar	6	10	16	12	8	52.00
9	Ku. E. U. Kotharkar	18	19	21	19	12	89.00
10	Ku. G. S. Kawle	AB	AB	AB	AB	AB	0.00
11	Ku. I. S. Borkar	12	16	18	13	11	70.00
12	Ku. J. S. Satone	AB	AB	AB	AB	AB	0.00
13	Ku. K. D. Bele	18	18	18	18	10	82.00
14	Ku. K. R. Raut	21	22	22	18	12	95.00
15	Ku. K. S. Thakare	20	22	19	21	11	93.00
16	Ku. L. S. Mahajan	15	18	21	20	11	85.00
17	Ku. M. G. Sayam	14	19	20	19	10	82.00
18	Ku. M. N. Hatwar	10	12	13	12	11	58.00
19	Ku. M. R. Verma	15	14	21	17	10	77.00
20	Ku. P. R. Umate	19	19	21	17	11	87.00
21	Ku. P. V. Lanjekar	6	10	13	10	8	47.00
22	Ku. R. K. Umathre	18	17	20	19	10	84.00
23	Ku. R. M. Shidodkar	19	19	21	19	12	90.00
24	Ku. R. P. Deshpande	AB	AB	AB	AB	AB	0.00



Dr. R. O. Ganjiwale
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha

25	Ku. R. S. Khurpade	11	18	18	14	10	71.00
26	Ku. R. S. Raut	7	13	18	12	10	60.00
27	Ku. R. V. Bhale	15	16	18	20	12	81.00
28	Ku. S. M. Yadav	20	20	21	19	10	90.00
29	Ku. S. N. Deshmukh	14	16	19	17	10	76.00
30	Ku. S. P. Tapre	8	7	13	9	9	46.00
31	Ku. S. R. Chahande	19	17	20	18	11	85.00
32	Ku. S. R. Thakare	14	15	18	15	11	73.00
33	Ku. S. R. Tiwari	12	16	17	15	11	71.00
34	Ku. S. S. Ahirrao	16	18	18	18	11	81.00
35	Ku. S. S. Misalkar	16	18	18	18	10	80.00
36	Ku. S. S. Nimbalkar	12	14	19	17	9	71.00
37	Ku. S. T. Yedlawar	15	17	18	15	10	75.00
38	Ku. S. V. Bhoyar	10	11	16	9	8	54.00
39	Ku. S. V. Dhok	14	16	21	18	10	79.00
40	Ku. Sakshi Raut	18	17	17	16	11	79.00
41	Ku. Shreya Raut	12	14	16	14	9	65.00
42	Ku. T. R. Dhage	18	16	19	18	11	82.00
43	Ku. T. R. Hukum	20	19	20	18	10	87.00
44	Ku. T. U. Moon	14	12	18	15	11	70.00
45	Ku. U. S. Gawande	AB	AB	AB	AB	AB	0.00
46	Ku. V. A. Kshirsagar	11	17	18	12	9	67.00
47	Ku. V. D. More	20	20	21	20	12	93.00
48	Ku. V. D. Shrivastava	21	21	20	22	11	95.00
49	Ku. V. N. Deshmukh	19	20	23	19	12	93.00
50	Ku. V. R. Sushir	16	17	20	17	11	81.00
51	Ku. Y. D. Kude	18	22	21	22	11	94.00
52	Mr. A. C. Darda	9	12	15	13	9	58.00
53	Mr. A. K. Kinhekar	8	8	13	12	9	50.00
54	Mr. A. N. Dahake	9	11	14	11	9	54.00
55	Mr. G. M. Sukhadia	12	13	19	15	9	68.00
56	Mr. H. C. Butale	13	13	15	14	12	67.00



(Dr. R. O. Ganjiwale)
Principal
PRINCIPAL
Institute of Pharmaceutical Education & Research
Borgaon (Mglic), Wardha

57	Mr. J. S. Kadu	8	4	12	9		7	40.00
58	Mr. M. A. Hydri	9	12	17	11		7	56.00
59	Mr. N. O. Dhanbhate	10	12	17	11		8	58.00
60	Mr. N. S. Chhajed	12	14	17	16		10	69.00
61	Mr. O. P. Patte	11	11	9	9		7	47.00
62	Mr. P. M. Zade	12	15	16	17		9	69.00
63	Mr. P. S. Waghe	12	14	17	12		11	66.00
64	Mr. P. S. Wanare	17	21	15	17		11	81.00
65	Mr. R. G. Bongade	17	21	19	18		11	86.00
66	Mr. R. G. Prabhakar	16	19	17	18		12	82.00
67	Mr. R. R. Magar	5	5	(16) 11	8		7	41.00
68	Mr. R. S. Dhumale	0	AB	AB	AB		AB	0.00
69	Mr. R. S. Jawade	9	10	15	14		7	55.00
70	Mr. S. B. Fulmalgi	10	12	15	12		10	59.00
71	Mr. S. C. Borkar	10	14	16	13		9	62.00
72	Mr. S. D. Dahare	19	20	19	19		11	88.00
73	Mr. S. H. Gawali	10	12	12	12		8	54.00
74	Mr. S. R. Bokade	7	5	10	8		7	37.00
75	Mr. T. M. Khade	16	15	17	15		10	73.00
76	Mr. Y. M. Karlekar	15	16	16	13		10	70.00
77	Mr. Y. R. Kondawar	11	10	14	11		8	54.00

Note : If any discrepancy is observed bring it to the notice of Examination In-Charge in writing before 10/11/2023 upto 2.00 p.m.
 Date : 09/11/2023

Entered By:-
 (Mr. V. A. Rekhate)

Checked By:-
 Ms. S. P. Gautam & Dr. B. R. Gandhare

(Dr. R. O. Ganjiwale)
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha



(Dr. R. O. Ganjiwale)
 Principal
PRINCIPAL
 Institute of Pharmaceutical Education & Research
 Borgaon (Meghe), Wardha